

Tetrahedron Letters 41 (2000) 155-159

TETRAHEDRON LETTERS

Regio- and stereoselective oxidative difunctionalization of alkylidene cyclohexenes

Sang-Phyo Hong, Matthias C. McIntosh,* Tosha Barclay and Wally Cordes [†]

Department of Chemistry and Biochemistry, University of Arkansas, Fayetteville, AR 72701, USA

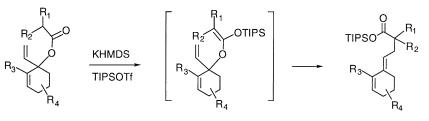
Received 10 June 1999; revised 15 October 1999; accepted 16 October 1999

Abstract

Stereo- and regioselective oxidation of alkylidene cyclohexenes affords novel γ -lactones, δ -lactones and related structures. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: dienes; lactones; lactonization; oxidation.

As part of a general program directed toward the synthesis of a variety of natural products, we have begun developing stereoselective approaches to alkylidene cycloalkenes.¹ For example, we have previously shown that Ireland Claisen rearrangement of bis-allyl silylketene acetals derived from readily accessible bis-allylic esters proceeds exclusively via the *exo* alkene to yield alkylidene cyclohexenes (Scheme 1).¹ We envisioned that oxidation of the diene could be used to install the requisite functionality of various natural product targets^{2–4} in a regio- and stereocontrolled manner.



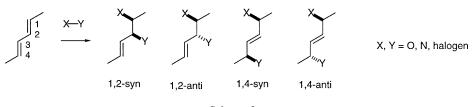
Scheme 1.

Oxidation of cyclic and acyclic dienes to afford 1,2- or 1,4-difunctionalized alkene adducts has been employed extensively in the preparation of a variety of synthetically useful building blocks.^{5–11} All possible modes of addition (1,2 and 1,4; *syn* and *anti*) have been observed depending upon the reagent(s) and reaction conditions (Scheme 2).^{5–11}

^{*} Corresponding author. E-mail: mcintosh@comp.uark.edu (M. C. McIntosh)

[†] Author to be contacted regarding X-ray crystallographic analyses.

^{0040-4039/00/\$ -} see front matter @ 1999 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(99)02043-2

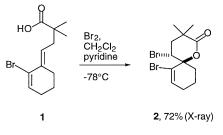


Scheme 2.

Oxidation involving the *exo* double bond of alkylidene cycloalkenes has the potential for control of the relative stereochemistry between the cycloalkene ring and the pendant side chain.¹² We report herein the realization of this approach to the synthesis of 1,2- and 1,4-difunctionalized cyclohexene lactones and related structures.

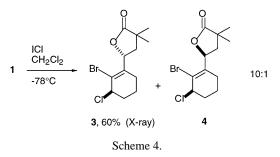
To simplify the initial studies of the oxidation chemistry, an achiral diene with electronically differentiated alkenes was employed. Reaction of various electrophilic oxidants with bromodiene 1^1 would be expected to occur at the more nucleophilic *exo* alkene (Scheme 3).¹³

Halolactonization of alkenoic acids is a well established method for the synthesis of lactones.¹⁴ Halolactonization of γ - δ enoic acids may afford either γ - or δ -lactones with poor to excellent selectivity depending upon the substitution pattern and reaction conditions.¹⁴ Halolactonization of γ , ϵ -dienic acids, however, has received very little attention.¹⁵ Bromolactonization of diene **1** afforded exclusively δ -lactone **2** via 6-endo cyclization (Scheme 3).¹⁶

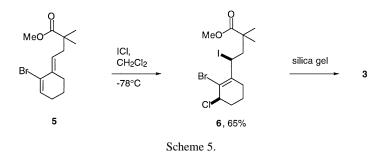


Scheme 3.

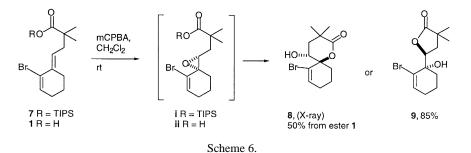
By contrast, treatment of diene **1** with ICl yielded solely γ -lactones **3** and **4** as a 10:1 mixture (Scheme 4). The stereochemistry of lactones **2** and **3** were determined unambiguously by X-ray crystallographic analysis.



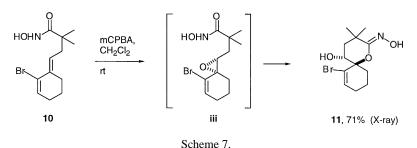
The regiochemical outcome of the chlorolactonization reaction apparently resulted from an initial regioselective and *syn*-stereoselective iodochlorination of the diene followed by cyclization via $S_N 2$ displacement of the iodide.^{9a,12a} This mechanism was supported by isolation of 1,4-*syn* iodochloride **6** upon treatment of dienic ester **5** under identical conditions (Scheme 5). The *syn* stereochemistry of iodochloride **6** was further supported by its partial lactonization to 1,4-*anti* chlorolactone **3** during silica gel chromatography.



Reaction of dienes **1** and **7**¹ with *m*-CPBA led initially to *exo* epoxides **i** and **ii**, respectively, based on ¹H NMR analysis of the reaction mixtures (Scheme 6). In situ lactonization occurred over 1 h at rt for acid **ii**, while TIPS ester **i** required ca. 16 h for complete cyclization. Dienic acid **1** yielded only γ -lactone **9** via 5-*exo* cyclization, whereas dienic TIPS ester **7** afforded only δ -lactone **8** via 6-endo cyclization.¹⁷ The structure of lactone **8** was verified by X-ray crystallographic analysis.

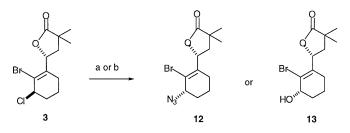


Interestingly, epoxidation of hydroxamic acid **10**, prepared from acid **1** via the corresponding acyl chloride, ultimately afforded (*Z*)-lactone oxime **11** in good yield (Scheme 7). Rapid formation of vinyl epoxide **iii** was followed by in situ cyclization to the oxime over 3 h at rt. This procedure may open a new entry to the synthesis of hydroxylated lactone oximes, which have previously been employed as 1,3-dipoles^{18a} and as intermediates in the synthesis of glycosylidene carbenes.^{18b}



We were also interested in using the oxidation chemistry for the stereoselective preparation of 1,4-diols and 1,4-aminoalcohols. Thus, chlorolactone **3** was converted to 1,4-*syn* azide **12** upon treatment with NaN₃ with 20:1 regioselectivity (ipso versus allylic substitution). Similarly, treatment of chlorolactone **3** with NaOAc and H₂O afforded 1,4-*syn* alcohol **13** with 10:1 regioselectivity (Scheme 8).

In summary, we have demonstrated that oxidation of alkylidene cyclohexenes can be employed in a systematic fashion to obtain 1,2- and 1,4-difunctionalized adducts. Use of the method in the synthesis of various natural products will be reported in due course.



Scheme 8. (a) NaN₃, DMF, rt, 67%; (b) NaOAc, DMF, H₂O, 120°C, 54%

Acknowledgements

Support was provided by the American Chemical Society (PRF-G 32150-G1), the Arkansas Science and Technology Authority (98-B-16) and NIH (R01-GM59406).

References

- 1. Zhang, X.; McIntosh, M. C. Tetrahedron Lett. 1998, 39, 7043-7046.
- For spirolide D, see: Hu, T.; Curtis, J. M.; Oshima, Y.; Quilliam, M. A.; Walter, J. A.; Watson-Wright, W. M.; Wright, J. L. C. J. Chem. Soc., Chem. Commun. 1995, 2159–2161.
- 3. For litophynol B, see: Miyamoto, T.; Yamada, K.; Ikeda, N.; Komori, T.; Higuchi, R. J. Nat. Prod. 1994, 57, 1212–1219.
- 4. For eupomatilone-6, see: Carroll, A. R.; Taylor, W. C. Aust. J. Chem. 1991, 44, 1705-1714.
- (a) For epoxidation, see: Jacobsen, E. N. J. Org. Chem. 1993, 58, 6939–6941; (b) Frohn, M.; Dalkiewicz, M.; Tu, Y.; Wang, Z.-X.; Shi, Y. J. Org. Chem. 1998, 63, 2948–2953; (c) Katsuki, T.; Martin, V. Org. React. 1996, 48, 1–299, and references cited therein.
- 6. For aziridination, see: Knight, J. G.; Muldowney, M. P. Synlett 1995, 949–951.
- 7. For dihydroxylation, see: Becker, H.; Soler, M. A.; Sharpless, K. B. Tetrahedron 1995, 51, 1345–1376.
- For Pd-catalyzed oxidation, see: Bäckvall, J.-E. In *The Chemistry of Dienes and Polyenes*; Rappoport, Z., Ed.; Wiley: New York, 1997; Vol. 1, pp. 653–681.
- For halogenation and co-halogenation, see: (a) Heasley, G. E.; Duke, M.; Hoyer, D.; Hunnicutt, J.; Lawrence, M.; Smolik, M. J.; Heasley, V. L.; Shellhamer, D. F. *Tetrahedron Lett.* **1982**, *23*, 1459–1462, and references cited therein; Han, X.; Khedakar, R. N.; Masnovi, J.; Baker, R. J. J. Org. Chem. **1999**, *64*, 5245–5250; (b) Barluenga, J.; Rodriguez, M. A.; Campos, P. J.; Asensio, G. J. Chem. Soc. Chem. Commun. **1987**, 1491–1492; Horiuchi, C. A.; Hosokawa, H.; Kanamori, M.; Muramatsu, Y.; Ochiai, K.; Takahashi, E. Chem. Lett. **1995**, 13–14; (c) for reviews, see: Rodriguez, J.; Dulcére, J.-P. Synthesis **1993**, 1177–1205; (d) Bahr, U.; von Brachel, H. In *Houben–Weyl Methoden der Organischen Chemie*; Thieme: Stuttgart, 1970; Vol. V/1c, pp 854-908. See also Ref. 12a.
- For anodic oxidation, see: Shono, T.; Kashimura, S. In *The Chemistry of Dienes and Polyenes*; Rappoport, Z., Ed.; Wiley: New York, 1997; Vol. 1, pp. 753–774.
- 11. For a general review of diene oxidation, see: Neumann, R.; Khenkin, A. M. In *The Chemistry of Dienes and Polyenes*; Rappoport, Z., Ed.; Wiley: New York, 1997; Vol. 1, pp. 889–926.
- For previous examples of oxidations of alkylidene cycloalkenes, see: (a) Ogawa, S.; Hattori, T.; Toyokuni, T.; Suami, T. *Bull. Chem. Soc. Jpn.* **1983**, *56*, 2077–2081; Boulin, B.; Arreguy-San Miguel, B.; Delmond, B. *Tetrahedron* **1998**, *54*, 2753–2762; (b) Wang, Z.-M.; Kakiuchi, K.; Sharpless, K. B. J. Org. Chem. **1994**, *59*, 6895–6897; Arjona, O.; León, M.; Plumet, J. J. Org. Chem. **1999**, *64*, 272–275; Chang, S.; Lee, N. H.; Hamada, T.; Irie, R.; Katsuki, T. Synlett **1994**, 479–481; (c) Myers, A. G.; Proteau, P. J.; Handel, T. M. J. Am. Chem. Soc. **1988**, *110*, 7212–7214.
- (a) Hudlicky, T.; Price, J. D.; Rulin, F.; Tsunoda, T. J. Am. Chem. Soc. 1990, 112, 9439–9440; Hudlicky, T.; Entwistle, D. A.; Pitzer, K. K.; Thorpe, A. J. Chem. Rev. 1996, 96, 1195–1220; (b) Kende, A. S.; Blacklock, T. J. Tetrahedron Lett. 1980, 21, 3119–3122.
- (a) For reviews, see: Dowle, M. D.; Davies, D. I. *Chem. Soc. Rev.* **1979**, *8*, 171-197; Bartlett, P. A. In *Asymmetric Synthesis*; Morrison, J. D., Ed.; Academic Press: Orlando, Florida, 1984; Vol. 3, pp 411–454; Cardillo, G.; Oreno, M. *Tetrahedron* **1990**, *46*, 3321–3408; (b) Snider, B. B.; Johnston, M. I. *Tetrahedron Lett.* **1985**, *26*, 5497–5500. See also Ref. 9c.
- Sato, A.; Ogiso, A.; Noguchi, H.; Mitsui, S.; Kaneko, I.; Shimada, Y. *Chem. Pharm. Bull.* **1980**, *28*, 1509–1525; Pearson, A. J.; Ray, T. *Tetrahedron* **1985**, *41*, 5765–5770.

- 16. All new compounds have been characterized by ¹H NMR, ¹³C NMR, MS, IR and combustion analysis.
- Control of the regiochemistry of epoxide opening by oxygen nucleophiles has been of considerable interest in recent years, largely resulting from synthetic approaches to the brevetoxins and other polyether toxins: Nicolaou, K. C.; Prasad, C. V. C.; Somers, P. K.; Hwant, C.-K. J. Am. Chem. Soc. 1989, 111, 5330–5334; Nicolaou, K. C.; Hwang, C.-K.; Duggan, M. E.; Nugiel, D. A.; Abe, Y.; Bal Reddy, K.; DeFrees, S. A.; Reddy, D. R.; Awartani, R. A.; Conley, S. R.; Rutjes, F. P. J. T.; Theodorakis, E. A. J. Am. Chem. Soc. 1995, 117, 10227–10238; Mori, Y.; Taegashi, K.; Furukawa, H. J. Am. Chem. Soc. 1997, 119, 4557–4558; Janda, K. D.; Shevlin, C. G.; Lerner, R. A. J. Am. Chem. Soc. 1995, 117, 2659–2660. For a review, see: Alvarez, E.; Candenas, M.-L.; Pérez, R.; Ravelo, J. L.; Martin, J. D. Chem. Rev. 1995, 95, 1953–1980. It is noteworthy that the regiochemistry of epoxide ring opening in this substrate was controlled by the structure of the oxygen nucleophile rather than the epoxide electrophile.
- (a) Yokoyama, M.; Yamada, N. *Tetrahedron Lett.* **1989**, *30*, 3675–3676; (b) Vasella, A.; Weber, M.; Textor, M.; Spencer, N. D. *Helv. Chim. Acta* **1998**, *81*, 1359–1372, and references cited therein.